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Tetrahedron Letters 47 (2006) 2323-2325

Tetrahedron Letters

Regioselective synthesis of *N*-acetylureas by manganese(III) acetate reaction of 1,3-disubstituted thioureas

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Received 11 January 2006; revised 31 January 2006; accepted 2 February 2006 Available online 20 February 2006

Abstract—Reactions of asymmetrical 1,3-disubstituted thioureas with manganese(III) acetate produce regioselective *N*-acetylureas. A mechanism for this novel transformation is proposed. © 2006 Elsevier Ltd. All rights reserved.

Since Bush and Finkbeiner's pioneering work reported three decades ago,¹ manganese(III)-based oxidative free radical reaction has become a powerful synthetic method.² One of the important applications of manganese(III) reagent is for acetylations. For example, $Mn(OAc)_3$ can be used for enantioselective acylation of α , β -unsaturated ketones³ and acetoxylation of fullerene derivatives.⁴ However, to the best of our knowledge, no such reagent has ever been used for N-acetylation reactions. We report here the first example of $Mn(OAc)_3$ -promoted regioselective N-acetylation of 1,3disubstituted thioureas in the synthesis of N-acetylureas.

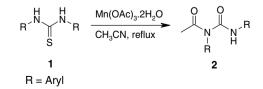
Ureas and thioureas are useful synthons for the construction of heterocyclic compounds.⁵ *N*-acylureas have important agrochemical⁶ and pharmaceutical applications.⁷ Dopamine D2 agonist Cabergoline, for example, has been used for the treatment of Parkinson's disease.⁸ For the synthesis of *N*-acylureas,⁹ direct acylation of symmetrical ureas¹⁰ or carbodiimides¹¹ is an efficient approach. However, because the reactions are not regioselective, only symmetrical ureas or carbodiimides produce single products.¹² The Mn(OAc)₃-mediated Nacetylation reactions described in this letter can be used to produce single *N*-acetylureas from symmetrical or asymmetrical thioureas bearing aryl and alkyl groups.

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We first attempted the acetylation using equimolar amount of $Mn(OAc)_3$ ·2H₂O and 1,3-di-*p*-tolylthiourea **1b** in MeCN. The 1-acetyl-1,3-di-*p*-tolylurea **2b** was generated in 10% yield. At a 2:1 molar ratio of $Mn(OAc)_3$ · 2H₂O to **1b**, the yield of **2b** was improved to 66% (Scheme 1). Reactions in different solvents (CH₂Cl₂, EtOH, MeOH, and AcOH) and at different temperatures were also attempted, but no further yield improvement was observed. The structure of compound **2b** (R = *p*-tolyl) was confirmed by X-ray crystallography analysis (Fig. 1).

Under the optimized conditions using 2 equiv of $Mn(OAc)_3$ ·2H₂O and MeCN as a solvent, reactions of symmetrical 1,3-diarylthioureas were performed and results are listed in Table 1.¹³ It was found that 1,3-diarylthioureas afforded N-acetylated products in good to excellent yields (Table 1, entries 1–10). In contrast, only trace amount of product was detected from reactions of 1,3-dialkylthioureas (Table 1, entries 11–12).

Asymmetrical 1,3-disubstituted thioureas 3^{14} were used to study the regioselectivity of acetylation reactions (Table 2). We found that reactions of



Scheme 1.

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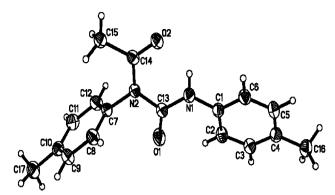
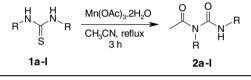


Figure 1. X-ray crystal structure of compound 2b.

Table 1. Acetylation of disubstituted thioureas 1a-l



Entry	Product	R	Yield ^a (%)
1	2a	Ph	72
2	2b	<i>p</i> -MeC ₆ H ₄	66
3	2c	o-MeC ₆ H ₄	63
4	2d	m-MeC ₆ H ₄	76
5	2e	<i>p</i> -MeOC ₆ H ₄	63
6	2f	p-ClC ₆ H ₄	86
7	2g	o-ClC ₆ H ₄	76
8	2h	p-BrC ₆ H ₄	88
9	2i	m-BrC ₆ H ₄	69
10	2j	$p-IC_6H_4$	52
11	2k	Benzyl	Trace
12	21	Cyclohexyl	Trace

^a After flash column chromatography.

1-aryl-3-alkylthioureas had similar or slightly lower yields than those of symmetrical 1,3-diarylthioureas, and only single regioisomer was isolated from reactions of asymmetrical 1,3-disubstituted thioureas. The structure of product **4e** was confirmed by X-ray crystallography analysis (Fig. 2).

A possible mechanism for the regioselective N-acetylation is proposed in Scheme 2. 1-Aryl-3-alkylthiourea **3** can exist as isothioureas **5** or **6**, in which **5** is more stable. Compound **5** reacts with Mn(OAc)₃ to produce **7**. The oxygen attack followed by the fragmentation of Mn(II) and release of H₂S¹⁵ produces **8**. This compound undergoes O \rightarrow N acyl migration¹⁶ to give *N*-acetylurea **4**. The aryl group is needed to promote the acetylation of **7**. This mechanism explains why yields from reactions of 1,3-diarylthioureas, and why it was difficult to acetylate 1,3-dialkylthioureas.

In summary, Mn(III)-mediated N-acetylation of 1,3disubstituted thioureas has been developed for regioselective synthesis of *N*-acetylureas. The reaction can be performed under mild conditions and give products in good yields. Table 2. Acetylation of asymmetrical disubstituted thioureas 3a-k



Entry	Product	\mathbb{R}^1	\mathbb{R}^2	Yield ^a (%)		
1	4a	Ph	Cyclohexyl	68		
2	4b	o-MeC ₆ H ₄	Cyclohexyl	60		
3	4c	<i>m</i> -MeC ₆ H ₄	Cyclohexyl	62		
4	4d	p-ClC ₆ H ₄	Cyclohexyl	73		
5	4 e	$p-IC_6H_4$	Cyclohexyl	64		
6	4f	Ph	Cyclopentyl	65		
7	4g	<i>p</i> -MeC ₆ H ₄	Cyclopentyl	59		
8	4h	m-MeC ₆ H ₄	Cyclopentyl	60		
9	4i	p-ClC ₆ H ₄	Cyclopentyl	71		
10	4j	p-IC ₆ H ₄	Cyclopentyl	64		
11	4k	Ph	Propyl	59		

^a After flash column chromatography.

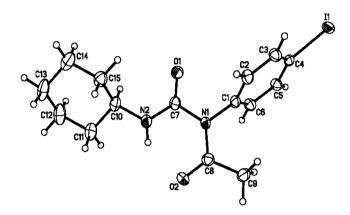
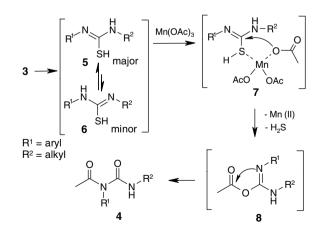


Figure 2. X-ray crystal structure of compound 4e.



Scheme 2. Proposed mechanism for N-acetylation.

Acknowledgements

We thank the Key Laboratory of Organic Synthesis of Jiangsu Province and Suzhou Scientific Committee, for financial supports (JSK016 and SG 0219).

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet. 2006.02.013.

References and notes

- Bush, J. B., Jr.; Finkbeiner, H. J. Am. Chem. Soc. 1968, 90, 5903; See also Williams, G. J.; Hunter, N. R. Can. J. Chem. 1976, 54, 3830.
- Reviews on manganese(III)-based oxidative free radical reaction: (a) Snider, B. B. Chem. Rev. 1996, 96, 339; (b) Melikyan, G. G. Synthesis 1993, 833; For recent examples of manganese(III)-based radical reactions, see: (c) Demir, A. S.; Reis, O.; Emrullahoglu, M. J. Org. Chem. 2003, 68, 578; (d) Zhang, Z.; Wang, G.-W.; Miao, C.-B.; Dong, Y.-W.; Shen, Y.-B. Chem. Commun. 2004, 1832; (e) Snider, B. B.; Duvall, J. R. Org. Lett. 2004, 6, 1265–1268; (f) Swain, N. A.; Brown, R. C. D.; Bruton, G. J. Org. Chem. 2004, 69, 122; (g) Huang, J. W.; Shi, M. J. Org. Chem. 2005, 70, 3859.
- Mn(OAc)₃-mediated acetoxylation of α,β-unsaturated ketones: (a) Demir, A. S.; Camkerten, N.; Gercek, Z.; Duygu, N.; Reis, O.; Arikan, E. *Tetrahedron* 1999, 55, 2441; (b) Demir, A. S.; Hamamci, H.; Doganel, F.; Camkerten, N.; Aksoy-Cam, H. *Turkish J. Chem.* 2000, 24, 141; (c) Demir, A. S.; Hamamcl, H.; Sesenoglu, O.; Aydogan, F.; Capanoglu, D.; Neslihanoglu, R. *Tetrahedron: Asymmetry* 2001, *12*, 1953; (d) Demir, A. S.; Sesenoglu, O. *Org. Lett.* 2002, *4*, 2021; (e) Tanyel, C.; Iyigun, C. *Tetrahedron* 2003, *59*, 7135; (f) Demir, A. S.; Reis, O.; Igdir, A. C. *Tetrahedron* 2004, *60*, 3427.
- 4. Chen, Z. X.; Wang, G. W. J. Org. Chem. 2005, 70, 2380.
- Zeng, R. S.; Zou, J. P.; Zhi, S. J.; Chen, J.; Shen, Q. Org. Lett. 2003, 5, 1657.
- (a) Wellinga, K.; Mulder, R.; van Daalen, J. J. J. Agric. Food Chem. 1973, 21, 348; (b) Nakagawa, Y.; Kitahara, K.; Nishioka, T.; Iwamura, H.; Fujita, T. Pestic. Biochem. Physiol. 1984, 21, 309.
- Goodman, L. S.; Gilman, A. The Pharmacological Basis of Therapeutics, 6th ed.; MacMillan: New York, 1980.

- 8. Brambilla, E.; Di Salle, E.; Briatico, G.; Mantegani, S.; Temperilli, A. Eur. J. Med. Chem. 1989, 24, 421.
- 9. Ravn, J.; Ankersen, M.; Begtrup, M.; Lau, J. F. Tetrahedron. Lett. 2003, 44, 6931, and references cited therein.
- (a) Weisz, I. Hung. Patent 49848; Chem. Abstr. 1990, 113, 171518; (b) Dmitrichenko, M. Y.; Rozinov, V. G.; Donskikh, V. I.; Ratovskii, G. V.; Sergienko, L. M.; Dolgushin, G. V.; Valeev, R. B. Zh. Obshch. Khim. 1988, 58, 2252–2261; (c) Yamada, K.; Kishikawa, K.; Kohmoto, S. Chem. Lett. 1988, 351; (d) Panzeri, A.; Ceriani, L.; Griggi, P.; Nesi, M. Eur. Patent 523668; Chem. Abstr. 1993, 118, 234337; (e) Panzeri, A.; Nesi, M.; DiSalle, E. WO Patent 9220700; Chem. Abstr. 1993, 118, 124875; (f) Maki, T.; Ishihara, K.; Yamamoto, H. Synlett 2004, 1355.
- (a) Corriu, R. J. P.; Gerard, F. L.; Perot-Petta, M. Synthesis 1991, 954; (b) Anglada, J. M.; Campos, T.; Camps, F.; Moreto, J. M.; Pages, L. J. Heterocycl. Chem. 1996, 33, 1259; (c) Zimmermann, D. M.; Olofson, R. A. Tetrahedron Lett. 1970, 11, 3453.
- Scott, W. A.; Kevin, E. H.; Andrew, M. A.; Peter, G. M. W. J. Org. Chem. 2002, 67, 7147.
- 13. General procedures for preparation of 1-acetyl-1,3-di-ptolylurea 2b: To a solution of 1,3-di-p-tolylthiourea 1b (0.26 g, 1 mmol) dissolved in 15 mL of acetonitrile was added Mn(OAc)₃·2H₂O (0.54 g, 2 mmol) in two portions. The solution was refluxed for 3 h until the color of solution changed from brown to pale vellow. After the removal of acetonitrile under vacuum, the residue was dissolved in acetone, filtered, concentrated and purified by flash column chromatography on silica gel and eluted with 4:1 petroleum ether/acetone to give 1-acetyl-1,3-di-ptolylurea 2b in 66% yield; mp 138-140 °C; IR (KBr, cm⁻¹) v 3211, 1713, 1663; ¹H NMR (CDCl₃): δ 11.35 (s, 1H, NH), 7.45–7.09 (m, 8H, $2C_6H_4$), 2.40 (s, 3H, CH₃), 2.29 (s, 3H, CH₃), 1.99 (s, 3H, CH₃); ¹³C NMR (CDCl₃): δ 175.5, 152.4, 139.5, 136.7, 135.6, 134.0, 130.8, 129.9, 129.1, 120.5, 27.0, 21.6, 21.3; HRMS (M^+) m/z calcd for C17H18N2O2 282.1368, found 282.1364.
- Ramadas, K.; Srinivasan, N.; Janarthanan, N. Tetrahedron Lett. 1993, 34, 6447.
- 15. H_2S was checked by $Pb(OAc)_2$ test paper.
- Volonterio, A.; Arellano, C. R.; Zanda, M. J. Org. Chem. 2005, 70, 2161.